Nitrates/Nitrites

Description/Chemical Forms:	Sources/Routes of Exposure:	Health Effects:
 NO₃⁻ (Nitrate): Naturally occurring inorganic ion found in soil, water, and some foods; common ingredient in nitrogen-based fertilizers, including ammonium or potassium nitrate NO₂⁻ (Nitrite): Naturally occurring inorganic ion found in soil, water, and some foods; can easily oxidize into nitrate Both NO₃⁻ and NO₂⁻ are part of the nitrogen cycle and a product of anaerobic digestion, in which microbes breakdown organic matter to produce ammonia gas (NH₃) 	Sources: foods (especially vegetables), compounds dissolved in soil or water Main Route of Exposure: Ingestion-dissolved compounds in both public and private wells from run-off or waste processing failures, residue left on produce, contaminated medication Inhalation-nitrite fumes can be released from some types of inhalants, such as "poppers"- volatiles that include cyclohexyl, butyl, and amyl nitrites used for sexual enhancement	Acute Acquired Methemoglobinemia: Excessive nitrates in the blood can alter iron levels from Fe²+ to Fe³+, causing the abnormal hemoglobin, methemoglobin, to form and restrict oxygen from reaching vital tissues • Early stages: bluish hue (cyanosis) to the skin, increased heart rate, fatigue • Advanced stages: CNS depression, lethargy, convulsions • Death can occur at blood methemoglobin levels ≥50%

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Prevention Strategies:

Diagnosis/Treatment Options:

No routine medical test to determine

Advise patients to:

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- Avoid direct contact with nitrogencontaining fertilizers.
- Regularly test private well water sources for nitrate levels; EPA current standard is 10mg/L or 10ppm

Sensitive Population (Fetal Methemoglobinemia): Infants under 4 months are especially sensitive to nitrate exposure since their GI tracts are highly alkaline and conducive to growth of nitrate-producing bacteria. Filtered water for formula is advisable for homes with high nitrogen content in private/domestic wells

More information concerning nitrate/nitrite exposure and health effects can be found at the following sites:

Links for Additional

Information:

http://www.atsdr.cdc.gov/toxfaqs/tf.asp ?id=1186&tid=258

http://water.epa.gov/drink/contaminant s/basicinformation/nitrate.cfm

http://www.drugabuse.gov/publications/drugfacts/inhalants



nitrite or nitrate levels in the blood currently exists. Methemoglobin can be detected in blood, yet causative factors, such as exposure to nitrite/nitrate, cannot be identified. These compounds have not officially been classified as carcinogenic to humans by the Department of Health and Human Services (DHH), the International Agency for Research on Cancer (IARC), or the Environmental Protection Agency (EPA), yet additional research will likely be necessary in the

future.